

Comparison study of CC and CH vibration frequencies and electronic properties for mono, Di, Tri, and tetra-rings layer of arm chair (SWCNTs)

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Abstract

Semi-empirical methods were applied for calculating the vibration frequencies and IR absorption intensities for normal coordinates of the {mono ($C_{56}H_{28}$), di ($C_{84}H_{28}$), tri ($C_{112}H_{28}$) and tetra ($C_{140}H_{28}$)} -rings layer for (7,7) armchair single wall carbon nanotube at their equilibrium geometries which were all found to have D_{7d} symmetry point group.

Assignment of the modes of vibration (3N-6) was done depending on the pictures of their modes by applying (Gaussian 03) program. Comparison of the vibration frequencies of (mono, di, tri and tetra) rings layer which are active in IR, and inactive in Raman spectra. For C-H stretching vibrations, the results showed that vibration frequencies value increased with increased of length nano tube (rings layer SWCNT).

The results include the relation for axial bonds, which are the vertical C-C bonds (annular bonds) in the rings and for circumferential bonds which are the outer ring bonds. Also include the assignment of puckering, breathing and clock-anticlockwise bending vibrations. They allow a comparative view of the charge density at the carbon atom too.

Key words

SWCNT single wall carbon nano tube, G03 aprogram for calculation, γ : Out of plane of the molecule, δ : In- plane of the molecule.

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دراسة مقارنة ترددات الاهتزاز لاواصر , CC, CH والخواص الكترونية (لاحادي وثنائي وثلاثي ورباعي) الطبقات الحلقية لانابيب ارم جير (لانابيب النانوكاربون أحادية الطبقة).

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الخلاصة

تم تطبيق طرائق الحسابات شبه التجريبية لحساب ترددات الاهتزاز وشدت امتصاص الأشعة تحت الحمراء لجميع الاحداثيات الداخلية لجزيئات) احادي ($C_{56}H_{28}$) وثنائي ($C_{84}H_{28}$) وثلاثي ($C_{112}H_{28}$) ورباعي ($C_{140}H_{28}$) الطبقات الحلقية لانابيب من نوع 7,7 arm chair لانابيب النانو كاربون على وفق الشكل الهندسي المتوازن والتي وجد امتلاكها للتماثل (D_{7d}) لها لمجموعة النقطة اعتمد حساب الانماط الاهتزازية وبعدها ((3N-6)) ومناقشتها تماثليا و تأصريا على وفق الشكل الهندسي التوازني لها باستعمال برنامج كاوس 2003. وتم مقارنة ترددات الانماط الاهتزازية لهذه الجزيئات الفعالة في طيف الأشعة تحت الحمراء IR والغير فعالة في طيف الرامان، بالنسبة لترددات مط الاصرة C-H وجد ان قيم ترددات الأنماط الاهتزازية تعتمد على، طول الانبوب الاحادي الطبقة (عدد الطبقات الحلقية) للجزيئات. النتائج تضمنت العلاقة بين اطوال الاواصر C-Caxial المتجه على طول المحور السيني كذلك اطوال اواصر C-C المتجه على طول محيط انبوب النانوكاربون أحادي الطبقة C-Cc. الطبقة كما تم وبدقة تعيين جميع أنماط الاهتزازية العائدة للحركات الانبعاجية و التنفسية والانحنائية باتجاه وعكس اتجاه عقرب الساعة، و التي تعود اليها التشوهات الحاصلة من جراء الاهتزاز، وتم حساب ودراسة توزيع الكثافة الالكترونية على ذرات الكاربون لهذه الجزيئات ايضا.

Introduction

Carbon nanotubes have been intensively studied due to their importance as building block in nanotechnology. The special geometry and unique properties of carbon nanotube offer great potential applications, including nanoelectronic devices, energy storage, gas sensing, chemical probe, electron transport, and biosensors, field emission display, etc. Such devices operate typically on the changes of electrical response characteristics of the nanowire active component with the application of an externally applied mechanical stress or the adsorption of chemical or bio-molecule.

The study of vibration of CNTs is for successful applications in nanotechnology. Specifically, some vibration modes of CNTs, e.g., radial breathing mode [1-4], beam-like bending mode, [5-6] and longitudinal mode [7], offers valuable probes for the molecular structures and the elastic properties of CNTs. On the other hand, CNTs consisting of straight concentric layers with circular cross-section could lose their structural symmetry due to the vibration in axial, circumferential and radial directions [8,9]. This could result in a sudden change in their physical properties (e.g., the electrical properties [10] and in turn, significantly affect their performance in nanostructures. Thus, similar to the buckling behaviour [11] the vibration of CNTs turns out to be a major topic of great interest in nanomechanics, considerable efforts [12] have been devoted to capturing the fundamental vibration behaviours of CNTs by using experimental techniques [1-2,6-7] and multi-scale modelling tools [3-4,8-9,13-15]. Recently, the interest of the mechanics of CNTs has been transferred from their fundamental behaviours to the effect of internal and external factors on the elastic properties [16-18], buckling [19-22,23] of CNTs. However no comparison

study could be found in the literature for a normal coordinate analysis of the simplest type of nanotube i.e. the (mono, di, tri and tetra) rings-layer CNTs.

Methods of calculation

Both G03 program by PM3 method [24], and Mopac 2000 program by PM3 method, Computational Packages have been employed throughout this study to compute the geometrical Parameters (bond lengths and bond angles) and the energetic properties [25] were applied throughout the present work.

Results and Discussion

To study the vibration motions of such molecules, one has to define its geometric parameters, and has to distinguish between the axial CC (C-Caxial) bonds and circumferential CC (C-Cc) bonds, Fig.1 shows the two types of bonds in an armchair CNT [26].

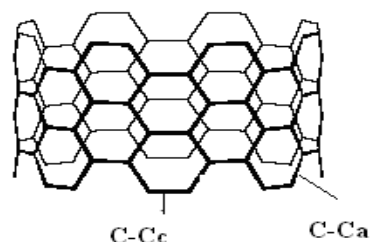


Fig.1: Structure of armchair CNT, indicating the two types of C-C bonds; C-Caxial (C-Ca) and C-Ccircumferential (C-Cc).

Various quantum mechanical studies were done for the physical properties of the nanotubes [27-31]. Structure deformation is expected to change their thermal and electronic properties too. Basic vibrations of SWCNTs were measured and assigned as breathing, puckering and clock-anti-clockwise deformation modes [32]. They are considered as finger print vibrations for the carbon nanotubes (CNTs) [33]. The active vibrations causes a change in its geometry structure Fig.2, The calculation gives different geometry were done to study the

impact of the puckering distortion on the electronic properties of CNTs [34-36]. For a normal mode of vibration to be infrared active, there must be a change in the dipole moment of the molecule during the course of vibration. During the vibration motion of a molecule, a regular fluctuation in the dipole

moment occurs, and a field is established which can interact with the electrical field associated with radiation. For the absorption of infrared radiation, a molecule must undergo a net change in its dipole moment as a result of its vibrational or rotational motions [37].

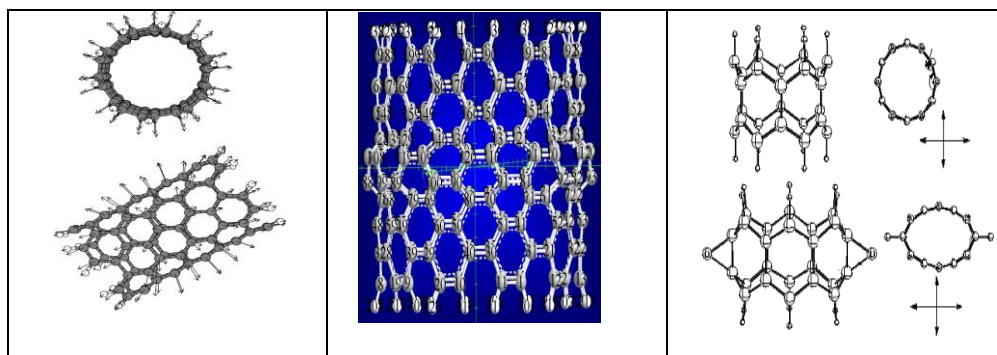


Fig.2: Structure deformation of a SWCNT as caused by its vibration motion.

The four classifications of carbon nanotubes (mono, di, tri and tetra)-rings layer, determined by the different of the numbers of rings and the length of CNT, can also be described as single-walled nanotubes (SWNT), resembling by rolling a graphene sheet into a cylinder mathematically structures are uniquely defined by specifying the coordinates of the smallest folding vector (n,n), (armchair) molecule is composed of

annulated number of member aromatic rings molecules. The (mono, di, tri and tetra)-rings layer SWCNTs are composed of annulated six member aromatic rings. Their (PM3) calculated equilibrium geometry shows D_{7d} symmetry [13-14], Fig. 3.

Due to their symmetry (D_{7d}) [38], Fig.4 and Table1, shows a repetitive sections of bonds. Table2 shows comparison of their calculated geometric parameters.

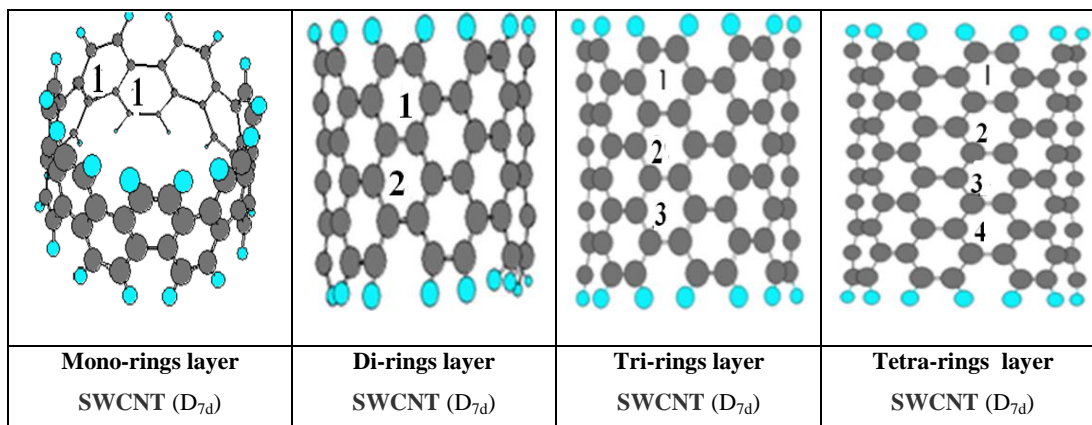


Fig.3: Equilibrium geometry for (mono, di, tri and tetra)-rings layer for (7,7) armchair single wall carbon nanotube (SWCNT) by using Mopac2000 program.

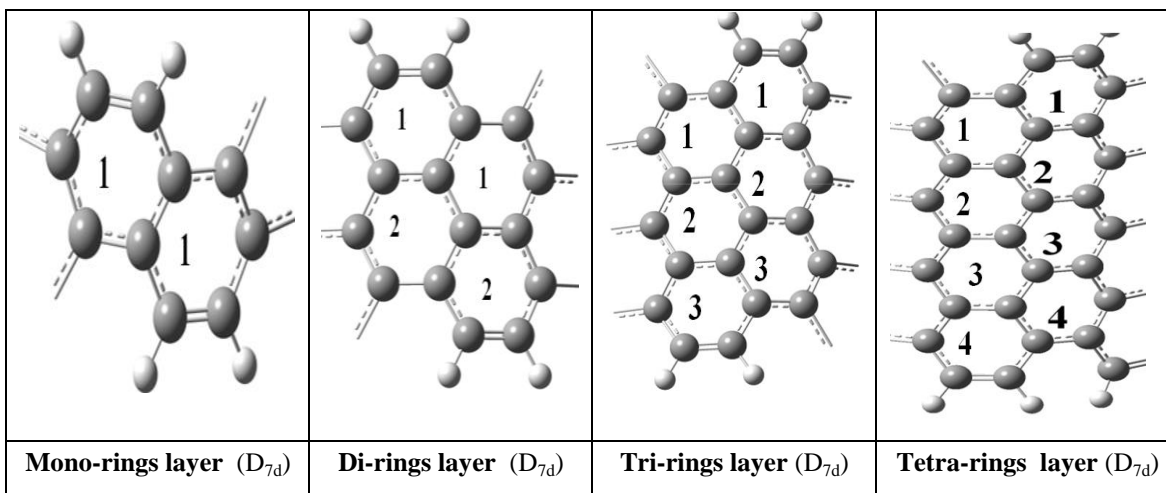


Fig.4: Repetitive sections of bonds and angles for (mono, di, tri and tetra) rings layer of (7,7) armchair SWCNT at their equilibrium geometries according to their point group (D_{7d}).

Table1: PM3 calculated bond distances of (mono, di, tri and tetra) rings layer for (7,7) Arm Chair SWCNT.

Arm Chair (SWCNT)	Diameter (Å)	CNT Length (Å)	Bond length (Å)						
			=C...Ca	—C...Ca	...C...Ca	C=Cc	C...Cc	C—Cc	C—H
Mono-ring layer D_{7d}	10.049	5.526	1.4302	-----	1.4005	1.3604	1.4351	-----	1.0997
Di-rings layer D_{7d}	10.047	7.974	1.4092	1.4065	1.4411	1.3802	1.4137	1.4521	1.0995
Tri-rings layer D_{7d}	10.103	10.444	1.4363	-----	1.4083 1.4224 1.4409	1.3583	1.4342 1.3975 1.4221	-----	1.0991
Tetra-rings layer D_{7d}	10.041	12.901	1.4246	-----	1.4031 1.4346 1.4301 1.4044	1.3670	1.4276 1.4292 1.3974 1.4395	-----	1.0996

C-Ca: axial bond.

C-Cc: circumferential bond

Noting that the C-Ca bonds in the angular SWCNTs are conjugated double bonds for all of different rings layer, but it may be connect to (CC double or single or conjugated double bond). The C-Cc bonds are double or conjugated double bond for (mono, tri and tetra)-rings layer SWCNT, and double or conjugated double bond or single bond for di-rings layer SWCNT.

The C=Cc bond decrease with increasing odd number or even numbers of rings layer. For odd rings layer mono and tri, (C-H) and

(C-Cc) bonds are longer in mono from that in tri, the reverse had noticed for even-rings layer di and tetra-rings layer.

Also note that (...C...Ca), increase in length from outer to centre in tri-rings layer, and decrease in length from outer to centre in tetra-rings layer. Fig.4, Table 1.

Table 2 shows some physical properties of calculated (mono, di, tri and tetra)-rings layer for (7,7) armchair SWCNT at their equilibrium geometry.

Table 2: Some physical properties of the calculated (mono, di, tri and tetra) rings layer for (7,7) armchair SWCNT at their equilibrium geometry by using Mopac2000 program.

Arm Chair (SWCNT)	M. Wt. g/ mol	ΔH_f kcal/mol,(kJ)	(debye) μ	HOMO (eV)	LUMO (eV)	ΔE (eV) LUMO-HOMO
Mono-ring layer $C_{56}H_{28}$	700.837	300.346 1256.648	0.000	-8.054	-1.213	6.841
Di-rings layer $C_{84}H_{28}$	1037.145	440.174 1841.689	0.000	-7.301	-2.206	5.095
Tri-rings layer $C_{112}H_{28}$	1373.453	601.062 2514.846	0.000	-7.008	-2.664	4.344
Tetra- rings layer $C_{140}H_{28}$	1709.761	721.486 3018.699	0.000	-7.403	-2.437	4.066

ΔH_f =heat of formation, HOMO=Highest Occupied molecular orbital, LUMO=Lowest Unoccupied molecular orbital, ΔE = The difference in energy levels.

ΔH_f , increase with increasing number of odd and even rings layer, ΔE (LUMO-HOMO) decrease with increasing number of rings layer that prove their electrical conductivity increased with increasing number of rings layer . Dipole moment μ is zero for all, because they all have center of inversion symmetry element. same relation noticed for vibrations modes (IR active).

Vibration frequencies assignment of Mono-rings layer ($C_{56}H_{28}$) armchair SWCNT

The di-rings layer SWCNT posses 246 fundamental vibrations (3N-6). Inspection of its irreducible representation, as defined by the symmetry (D_{7d}) character table, results in the following modes of vibration;

$$\Gamma_{\text{vibration}} = \Gamma_{\text{total}} - (\Gamma_{\text{rotation}} + \Gamma_{\text{translation}}) = 3N - 6 = 246$$

$$\Gamma_{\text{vibration}} = 18E_{3u} + 18E_{3g} + 18E_{2g} + 18E_{2u} + 17E_{1u} + 17E_{1g} + 9A_{1u} + 8A_{2g} + 9A_{1g} + 8A_{2u} = 246$$

There are 42 vibration modes (IR active) ($17E_{1u}$, and $8A_{2u}$). Their assignments are as follows:

CH stretching vibration

These are 28 in number, the displacement vectors of which are located at the H atoms.

Their frequency values range from ($3005.2-3029.4 \text{ cm}^{-1}$) (IR active), Inspecting the values in Table 3.

Ring (CCC) stretching vibration

Their displacement vectors are not located at definite C atoms as could be seen from the atomic displacement vectors. Their calculated vibration frequencies (IR active) as modes range from ($1347.6-1809.5 \text{ cm}^{-1}$), Inspecting the values in Table 3.

In-plane CH bending vibration (δCH)

There are as few as the C-H bonds. Their displacement vectors are mainly located at the corresponding H atoms. Their calculated vibration frequencies (IR active) as modes range from ($237.5-1105.6 \text{ cm}^{-1}$),. Inspecting the values in Table 3.

In-plane (CCC) bending vibration (δCCC)

Their calculated vibration frequencies (IR active) as modes range from ($326.4-1011.7 \text{ cm}^{-1}$), Inspecting the values in Table 3.

Out of plane CH bending vibration (γCH)

Their calculated vibration frequencies (IR active) as modes range from ($572.8-1215.4 \text{ cm}^{-1}$), Inspecting the values in Table3.

Out of plane ring bending vibrations (γ CCC)

The modes include puckering deformations, as well as breathing vibrations of the whole

ring. Their calculated vibration frequencies (IR-active) as modes range from (572.8-1215.4 cm^{-1}), Inspecting the values in Table3.

Table 3: Vibration frequencies and IR absorption intensities for (mono-rings layer ($C_{56}H_{28}$) molecule as calculated applying G03 program.

EU	Symmetry & description	PM3 Freq. cm^{-1}	Intensity km/mol
v ₂₃₇	CH str.	3029.383	28.714
v ₂₂₇	CH str.	3005.239	107.300
v ₂₁₅	C=C str	1809.491	15.036
v ₁₉₁	ring)CC _{str.} ((elongation)	1625.087	0.0384
v ₁₈₇	C--C str	1593.311	12.2663
v ₁₇₉	C=C str	1526.865	17.4909
v ₁₅₃	γ CH(wag.)	1197.14	16.559
v ₁₃₆	δ CH(siccer.)	1105.650	0.4390
v ₁₂₅	δ CH(rock))+ δ ring (δ CCC)	1011.752	0.2571
v ₁₀₃	γ CH (wagg.)+ γ ring (γ CCC)	890.496	18.575
v ₉₄	γ CH (wagg.)+ γ ring (γ CCC)	852.991	93.477
v ₈₆	δ CH(rock)	812.674	137.946
v ₇₄	γ CH (wagg)+ γ ring (γ CCC)	702.001	1.791
v ₆₀	γ CH (wagg)+ γ ring (γ CCC)	572.863	13.074
v ₃₄	δ CH(rock)	388.437	15.734
v ₂₈	δ CH(rock)+ δ ring (δ CCC)	326.432	11.190
v ₂₁	δ CH(rock)	237.476	10.236
A_{2u}			
v ₂₃₁	CH str.	3010.232	358.860
v ₁₉₁	C=C str	1695.497	8.1662
v ₁₆₅	C=C str	1347.570	7.2816
v ₁₅₆	γ CH (wagg)+ γ ring (γ CCC)	1215.372	48.906
v ₁₃₂	δ CH(siccer.)	1077.490	0.6641
v ₈₈	δ CH(rock))+ δ ring (δ CCC)	814.134	0.590
v ₅₁	δ CH(rock))+ δ ring (δ CCC)	516.299	7.919

Vibration frequencies assignment of di-rings layer ($C_{84}H_{28}$) armchair SWCNT

The di-rings layer SWCNT posses 433 fundamental vibrations (3N-6). Inspection of its irreducible representation, as defined by the symmetry (D_{7d}) character table, results in the following modes of vibration;

$$\Gamma_{\text{vibration}} = \Gamma_{\text{total}} - (\Gamma_{\text{rotation}} + \Gamma_{\text{translation}}) = 3N - 6 = 330$$

$$\Gamma_{\text{vibration}} = 24E_{3u} + 24E_{3g} + 24E_{2g} + 24E_{2u} + 23E_{1u} + 23E_{1g} + 12A_{1u} + 11A_{2g} + 12A_{1g} + 11A_{2u} = 330$$

There are 59 vibration modes (IR active) ($24E_{2u}$, and $11A_{2u}$). Their assignments are as follows:

CH stretching vibration

These are 28 in number, the displacement vectors of which are located at the H atoms. Their frequency values as modes range from (3004.4- 3029.3 cm^{-1}) (IR active), Inspecting the values in Table 4.

Ring (CCC) stretching vibration

Their displacement vectors are not located at definite C atoms as could be seen

from the atomic displacement vectors. Their calculated vibration frequencies (IR active) as modes range from (1242.7-1786.93 cm^{-1}),. Inspecting the values in Table 4.

In-plane CH bending vibration (δCH)

There are as few as the C-H bonds. Their displacement vectors are mainly located at the corresponding H atoms. Their calculated vibration frequencies (IR active) as modes range from (161.6 -1001.6 cm^{-1}), Inspecting the values in Table 4.

In-plane (CCC) bending vibration (δCCC)

symmetric modes range from (149.6-914.9 cm^{-1}), Inspecting the values in Table4.

Out of plane CH bending vibration (γCH)

Their calculated vibration frequencies (IR active) as modes range from (348.6-1284.0 cm^{-1}), inspecting the values in table 4.

Out of plane ring bending vibrations (γCCC)

The modes include puckering deformations, as well as breathing vibrations of the whole ring. Their calculated vibration frequencies (IR-active) as modes range from (321.1-1284.0 cm^{-1}), Inspecting the values in Table4.

Vibration frequencies assignment of tri-rings layer ($\text{C}_{1\downarrow\uparrow}\text{H}_{28}$) armchair

The tri-rings layer SWCNT posses 414 fundamental vibrations (3N-6). Inspection of its irreducible representation, as defined by the symmetry character table, results in the following modes of vibration;

$$\Gamma_{\text{vibration}} = \Gamma_{\text{total}} - (\Gamma_{\text{rotation}} + \Gamma_{\text{translation}}) = 3N - 6 = 414$$

$$\Gamma_{\text{vibration}} = 30\text{E}_3\text{u} + 30\text{E}_3\text{g} + 30\text{E}_2\text{g} + 30\text{E}_2\text{u} + 29\text{E}_1\text{u} + 29\text{E}_1\text{g} + 15\text{A}_1\text{u} + 14\text{A}_2\text{g} + 15\text{A}_1\text{g} + 14\text{A}_2\text{u} = 414$$

There are 72 vibration modes (IR active)

(29 E_1u , and 14 A_2u). Their assignments are as follows:

CH stretching vibration

These are 28 in number, the displacement vectors of which are located at the H atoms. Their frequency values as modes range from (3008.1- 3032.1 cm^{-1}) (IR active), Inspecting the values in Table 5.

Ring (CCC) stretching vibration

Their displacement vectors are not located at definite C atoms as could be seen from the atomic displacement vectors. Their calculated vibration frequencies (IR active) as modes range from (1416.9- 1796.6 cm^{-1}), Inspecting the values in Table 5.

In-plane CH bending vibration (δCH)

There are as few as the C-H bonds. Their displacement vectors are mainly located at the corresponding H atoms. Their calculated vibration frequencies (IR active) as modes range from (138.7- 792.1 cm^{-1}), Inspecting the values in Table 5.

In-plane bending (CCC) vibration (δCCC)

Their calculated vibration frequencies (IR active) as modes range from (650.3- 1116.8 cm^{-1}), Inspecting the values in Table 5.

Out of plane CH bending vibration (γCH)

Their calculated vibration frequencies (IR active) as modes range from (278.4- 1364.1 cm^{-1}), Inspecting the values in Table 5.

Out of plane ring bending vibration (γCCC)

The modes include puckering deformations, as well as breathing vibrations of the whole ring. Their calculated vibration frequencies (IR active) as modes range from (278.4-1364.1 cm^{-1}),. Inspecting the values in Table5.

Table 4: Vibration frequencies and IR absorption intensities for (di-rings layer ($C_{84}H_{28}$) molecule as calculated applying G03 program by PM3 method.

EU	Symmetry & description	PM3 Freq. cm ⁻¹	Intensity km/mol
v ₃₂₁	CH str.	3029.32	5.157
v ₃₁₃	CH str.	3004.43	87.929
v ₂₉₀	C=C str	1786.93	29.090
v ₂₈₆	ring (CC _{str.})	1765.19	0.033
v ₂₅₉	C- C str	1624.74	1.608
v ₂₅₁	C- C str	1590.6	14.307
v ₂₃₈	C=C str	1478.47	22.973
v ₂₃₀	C- C str	1434.80	25.709
v ₂₁₃	γCH(twiss.)+γring (γCCC)	1284.01	21.413
v ₁₉₃	γCH (wag.)	1135.610	1.084
v ₁₇₈	γCH(siccer.)	1058.010	3.844
v ₁₆₉	δCH (rock.)	1001.142	11.930
v ₁₅₃	γring (γCCC)+γCH(twiss.)	948.942	0.5129
v ₁₅₀	δCH+δring (δCCC)(siccer)	914.984	30.582
v ₁₂₉	δCH(rock.)	819.102	207.138
v ₉₉	γring (γCCC)+γCH(twiss.)	672.731	0.0734
v ₈₇	γCH (wagg)+ γring (γCCC)	630.422	0.199
v ₈₀	γring (γCCC)+γCH(twiss.)	582.769	2.0624
v ₇₄	γCH (wagg)+ γring (γCCC)	530.382	47.086
v ₆₃	γring (γCCC)+γCH (wag.)	480.462	0.0103
v ₄₁	γCH(twiss.)+ γring (γCCC)	348.624	0.348
v ₃₅	γring (γCCC)breath.	321.064	4.169
v ₁₅	δCH (rock.)	161.639	8.1066
A _{2u}			
v ₃₁₅	CH str.	3008.51	484.441
v ₂₇₉	ring)CCstr.(elong.)	1746.01	4.6279
v ₂₄₈	C=C str	1569.93	5.1196
v ₂₃₇	C- C str	1469.77	56.657
v ₂₀₇	ring)CCstr.(elong.)	1242.74	0.0583
v ₁₈₀	γCH(twist.)	1064.96	9.1125
v ₁₅₅	γCH(twist.)+γring (γCCC)	955.296	5.856
v ₁₂₃	δCH(rock.)	797.23	1.8236
v ₈₆	γCH (wagg)+ γring (γCCC)	611.612	4.1519
v ₄₅	δCH+δring (δCCC) breath.	358.916	1.4007
v ₁₀	δCH+δring (δCCC)	149.602	0.376

Table 5: Vibration frequencies and IR absorption intensities for (di-rings layer ($C_{112}H_{28}$) molecule as calculated applying G03 program by PM3 method.

EU	Symmetry & description	PM3 Freq. cm ⁻¹	Intensity km/mol
V ₄₀₅	CH str.	3032.11	6.8021
V ₃₉₇	CH str.	3008.1	142.728
V ₃₇₇	C- C str	1796.58	10.2513
V ₃₇₁	C- C str	1784.38	0.0041
V ₃₄₉	C=C str	1710.91	0.626
V ₃₂₈	C- C str	1624.18	0.348
V ₃₁₂	C=C str	1574.96	14.8256
V ₃₀₁	ring (CCC _{str.})	1545.87	15.710
V ₂₉₂	ring (CCC _{str.})	1466.53	5.0778
V ₂₈₁	ring (CCC _{str.})	1416.97	0.5857
V ₂₇₈	γ CH(wagg.)+ γ ring (γ CCC)	1364.08	1.2167
V ₂₄₇	γ CH(wagg.)+ γ ring (γ CCC)	1155.24	11.7072
V ₂₃₄	γ CH(wagg.)+ γ ring (γ CCC)	1080.82	0.3068
V ₂₂₁	γ CH(twiss.)	1036.62	3.7063
V ₂₁₄	γ CH(twiss.)	998.785	0.5345
V ₁₉₇	γ CH(twiss.)+ γ ring (γ CCC)	937.036	11.6353
V ₁₈₈	γ CH(twist.)+ γ ring (γ CCC)	899.016	15.649
V ₁₆₆	γ CH(wagg.)+ γ ring (γ CCC)	824.343	32.543
V ₁₆₀	γ CH(wagg.)+ γ ring (γ CCC)	806.363	130.246
V ₁₄₄	γ CH(wagg.)+ γ ring (γ CCC)	747.612	2.318
V ₁₀₉	γ CH(wagg.)+ γ ring (γ CCC)	623.098	45.345
V ₁₀₇	γ CH(twiss.)+ γ ring (γ CCC)	611.811	0.1942
V ₉₃	γ ring (γ CCC)+ γ CH(twiss.)	540.388	1.5285
V ₈₇	γ CH(twiss.)+ γ ring (γ CCC)	505.551	0.3273
V ₆₅	γ CH(wagg.)+ γ ring (γ CCC)	395.983	13.2551
V ₅₇	γ CH(twiss.)+ γ ring (γ CCC)	361.336	2.526
V ₄₀	γ CH(twiss.)+ γ ring (γ CCC)	318.444	6.3217
V ₃₈	γ ring (γ CCC)	306.600	0.0033
V ₁₁	δ CH(rock.)	144.079	9.707
A _{2u}			
V ₃₉₉	CH str.	3013.16	497.47
V ₃₆₃	C- C str	1766.76	9.276
V ₃₄₅	C=C str	1690.55	4.9801
V ₃₁₈	ring (CCC _{str.})	1588.24	0.5889
V ₂₉₉	γ ring (γ CCC)	1531.39	0.4049
V ₂₆₃	γ CH(twist.)+ γ ring (γ CCC)	1267.74	91.423
V ₂₃₈	δ ring (δ CCC)	1116.81	1.2855
V ₂₂₀	γ CH(twist.)	1028.42	0.2526
V ₁₅₅	δ CH+ δ ring (δ CCC)	792.111	3.3319
V ₁₅₁	δ CH+ δ ring (δ CCC)	771.932	8.8077
V ₁₂₃	δ CH(rock.)+ δ ring (δ CCC)	650.264	0.4184
V ₇₀	γ CH(wagg.)+ γ ring (γ CCC)	428.731	4.944
V ₃₅	γ CH(wagg.)+ γ ring (γ CCC)	278.406	0.7373
V ₁₀	δ CH(rock.)	138.721	0.2038

Vibration frequencies assignment of tetra-rings layer (C₁₄₀H₂₈) armchair SWCNT.

The tetra-rings layer possesses 498 fundamental vibrations (3N-6). Inspection of its irreducible representation, as defined by the symmetry character table, results in the following modes of vibration;

$$\Gamma_{\text{vibration}} = \Gamma_{\text{total}} - (\Gamma_{\text{rotation}} + \Gamma_{\text{translation}}) = 3N - 6 = 498$$

$$\Gamma_{\text{vibration}} = 36E_{3u} + 36E_{3g} + 36E_{2g} + 36E_{2u} + 35E_{1u} + 35E_{1g} + 18A_{1u} + 17A_{2g} + 18A_{1g} + 17A_{2u} = 498$$

These are 87 IR active (35E_{1u}, and 17A_{2u}). Their assignments are as follow:

Relative to the σ_h reflection the vibration modes are classified as symmetric and antisymmetric modes.

CH stretching vibration

These are 28 in number, the displacement vectors of which are located at the H atoms. Their calculated vibration frequencies (IR active) as modes range from (3004.3-3029.0 cm⁻¹), Inspecting the values in Table 6.

Ring (CCC) stretching vibration

Their displacement vectors are not located at definite C atoms as could be seen from the atomic displacement vectors. Their calculated vibration frequencies (IR active) as modes range from (1409.13-1796.6 cm⁻¹), Inspecting the values in Table 6.

In-plane CH bending vibration (δ CH)

There are as few as the C-H bonds. Their displacement vectors are mainly located at the corresponding H atoms. Their calculated vibration frequencies (IR active) as modes range from (149.7- 1067.7 cm⁻¹), Inspecting the values in Table 6.

In-plane (CCC) bending vibration (δ CCC)

Their calculated vibration frequencies (IR active) as modes range from (149.7- 1343.7 cm⁻¹), Inspecting the values in Table 6.

Out of plane CH bending vibration (γ CH)

Their calculated vibration frequencies (IR active) as modes range from (270.1- 1288.6 cm⁻¹), Inspecting the values in Table 6.

Out of plane ring bending vibration (γ CCC)

The modes include puckering deformations, as well as breathing vibrations of the whole ring. Their calculated vibration frequencies (IR active) as modes range from (256.3-1363.9 cm⁻¹), Inspecting the values in Table 6.

Table 6, includes the calculated frequencies as calculated applying semi-empirical methods (PM3) by Gaussian03 program and IR absorption intensities of the vibration modes. Commonly known, the vibrations with (intensity= 0.0) are forbidden (IR inactive, Raman active), those with (intensity \neq 0.0) are allowed (IR active, Raman inactive).

Fig. 5 shows the vibration pictures for some modes of the (7,7) CNT (Armchair), as calculated applying semi-empirical methods (PM3) by Gaussian 03 program.

The whole relations

$\nu_{\text{CH}_{\text{str.}}(\text{tri-rings layer})} > \nu_{\text{CH}_{\text{str.}}(\text{mono-rings layer})} > \nu_{\text{CH}_{\text{str.}}(\text{di-rings layer})} > \nu_{\text{CH}_{\text{str.}}(\text{tetra-rings layer})}$

$\nu_{\text{CC}_{\text{str.}}(\text{mono-rings layer})} > \nu_{\text{CC}_{\text{str.}}(\text{tri=tetra-rings SWCNT})} > \nu_{\text{CC}_{\text{str.}}(\text{di-rings layer})}$

$\nu_{\delta\text{ring}(\text{tetra-rings layer})} > \nu_{\delta\text{ring}(\text{tri-rings layer})} > \nu_{\delta\text{ring}(\text{mono-rings layer})} > \nu_{\delta\text{ring}(\text{di-rings layer})}$

$\nu_{\gamma\text{ring}(\text{tri-rings layer})} > \nu_{\gamma\text{ring}(\text{tri-rings layer})} > \nu_{\gamma\text{ring}(\text{di-rings layer})} > \nu_{\gamma\text{ring}(\text{mono-rings layer})}$

$\nu_{\delta\text{CH}(\text{mono-rings layer})} > \nu_{\delta\text{CH}(\text{tetra-rings layer})} > \nu_{\delta\text{CH}(\text{di-rings layer})} > \nu_{\delta\text{CH}(\text{tri-rings layer})}$

$\nu_{\gamma} \text{CH}(\text{tri-rings layer}) > \nu_{\gamma} \text{CH}(\text{tetra-rings layer}) > \nu_{\gamma} \text{CH}(\text{di-rings layer}) > \nu_{\gamma} \text{CH}(\text{mono-rings layer})$

The results include the relation for odd ring number in length.

$\nu_{\text{CH}_{\text{str.}}}(\text{tri-rings layer}) > \nu_{\text{CH}_{\text{str.}}}(\text{mono-rings layer})$

$\nu_{\text{CC}_{\text{str.}}}(\text{mono-rings layer}) > \nu_{\text{CC}_{\text{str.}}}(\text{tri=tetra-rings})$

$\nu_{\delta\text{ring}}(\text{tri-rings layer}) > \nu_{\delta\text{ring}}(\text{mono-rings layer})$

$\nu_{\delta\text{CH}}(\text{mono-rings layer}) > \nu_{\delta\text{CH}}(\text{tri-rings layer})$

$\nu_{\gamma\text{ring}}(\text{tri-rings layer}) > \nu_{\gamma\text{ring}}(\text{mono-rings layer})$

$\nu_{\gamma} \text{CH}(\text{tri-rings layer}) > \nu_{\gamma} \text{CH}(\text{mono-rings layer})$

The results include the relation for even ring number in length.

$\nu_{\text{CH}_{\text{str.}}}(\text{di-rings layer}) \approx \nu_{\text{CH}_{\text{str.}}}(\text{tetra-rings layer})$

$\nu_{\text{CC}_{\text{str.}}}(\text{tetra-rings layer}) > \nu_{\text{CC}_{\text{str.}}}(\text{di-rings})$

$\nu_{\delta\text{ring}}(\text{tetra-rings layer}) > \nu_{\delta\text{ring}}(\text{di-rings layer})$

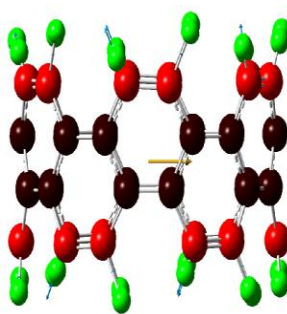
$\nu_{\delta\text{CH}}(\text{tetra-rings layer}) > \nu_{\delta\text{CH}}(\text{di-rings layer})$

$\nu_{\gamma\text{ring}}(\text{tetra-rings layer}) > \nu_{\gamma\text{ring}}(\text{di-rings layer})$

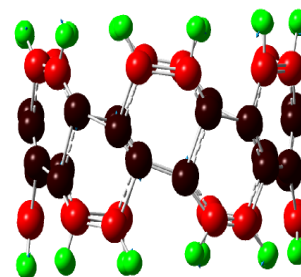
Finally, the calculated vibration frequencies (IR-active) show that vibration frequency values of (C-C stretching), (γ_{ring} , $\nu_{\delta\text{ring}}$) Vibrations frequency shows higher frequency values with increased numbers of rings layer in length lead to many more electronic and mechanical applications.



asym. (γCCC) (puck.)



asym. CH str.



(sym. ring) C--C str.

Fig.5: The graphical pictures of some vibration modes for CNT (armchair) as calculated applying G03 program.

Table 6: Vibration frequencies and IR absorption intensities for (tetra-rings layer ($C_{140}H_{28}$) molecule as calculated applying G03 program.

EU	Symmetry & description	PM3 Freq. cm^{-1}	Intensity km/mol
V ₄₈₉	CH str.	3029.03	9.0755
V ₄₈₁	CH str.	3004.26	117.43
V ₄₆₁	C-C str	1796.56	4.349
V ₄₅₇	C=C str	1793.22	14.884
V ₄₃₅	C=C str	1722.35	2.539
V ₄₁₄	C=C str	1675.17	0.278
V ₄₀₃	C-C str	1643.15	1.444
V ₃₉₉	ring (CCC _{str.})	1602.76	0.0197
V ₃₆₉	ring (CCC _{str.})	1556.18	0.4601
V ₃₆₃	C- C str	1536.56	8.336
V ₃₅₂	ring (CCC _{str.})	1498.85	17.8001
V ₃₃₄	ring (CCC _{str.})	1409.13	30.762
V ₃₂₁	δ ring (δ CCC)	1343.71	0.2808
V ₃₁₃	γ CH(wagg.)+ γ ring (γ CCC)	1288.64	6.322
V ₂₉₅	γ CH(wagg.)	1152.63	5.5834
V ₂₇₄	δ CH(siccer.)	1067.71	1.0663
V ₂₆₀	δ CH(siccer.)	1004.26	0.076
V ₂₄₇	γ ring (γ CCC)+ γ CH	964.564	1.2371
V ₂₄₁	γ CH(twiss.)+ γ ring (γ CCC)	945.735	0.0122
V ₂₂₇	γ CH(twiss.)+ γ ring (γ CCC)	896.6	5.1302
V ₂₁₆	δ CH+ δ ring (δ CCC)	858.66	22.623
V ₁₉₄	δ CH(rock)	812.82	125.337
V ₁₅₃	γ CH(twiss.)+ γ ring (γ CCC)	674.771	1.1662
V ₁₅₀	δ CH+ δ ring (δ CCC)	664.392	11.594
V ₁₄₃	δ CH(rock.)+ + δ ring (δ CCC)	653.804	7.2941
V ₁₂₄	γ CH(wagg.)+ γ ring (γ CCC)	602.856	3.0915
V ₉₉	γ CH(twiss.)+ γ ring (γ CCC)	492.552	22.087
V ₉₅	γ CH(twiss.)+ γ ring (γ CCC)	481.314	1.5762
V ₈₂	γ CH(twiss.)+ γ ring (γ CCC)	436.045	4.3508
V ₆₁	γ CH(twiss.)+ γ ring (γ CCC)	357.166	2.9102
V ₅₉	γ CH(twiss.)+ γ ring (γ CCC)	346.147	1.0289
V ₄₄	γ ring (γ CCC)+ γ CH(wagg.)	284.641	0.935
V ₃₈	γ CH(wagg.)+ γ ring (γ CCC)	270.122	5.758
V ₁₃	δ CH(rock.)+ δ ring (δ CCC)	149.751	6.048
A _{2u}			
V ₄₈₃	CH str.	3008.892	616.385
V ₄₄₂	C=C str	1765.63	14.298
V ₄₀₉	C=C str	1654.39	3.1335
V ₃₈₇	ring (CCC _{str.}) elonga.	1600.18	4.349
V ₃₆₇	ring (CCC _{str.}) elonga.	1542.98	2.304
V ₃₄₈	ring (CCC _{str.})elonga.	1480.57	0.2087
V ₃₂₉	γ ring (γ CCC)	1363.96	100.904
V ₃₀₃	ring (CCC _{str.})	1193.15	5.645
V ₂₈₄	γ CH(twist.)	1114.75	20.534
V ₂₆₀	γ CH(twist.)	1028.51	1.906
V ₁₈₈	δ CH(rock)	801.216	4.297
V ₁₄₇	γ ring (γ CCC)+ γ CH(breath.)	660.794	0.366
V ₁₃₈	γ ring (γ CCC)+ γ CH(breath.)	642.957	2.8819
V ₁₀₉	δ CH(rock.)+ δ ring (δ CCC)	518.928	10.4807
V ₆₇	δ CH(rock.)+ δ ring (δ CCC)	376.334	1.735
V ₃₅	γ ring (γ CCC)(breath.)	256.287	0.4318
V ₁₅	δ CH(rock.)+ δ ring (δ CCC)	152.02	0.7306

Table 7: Calculated Vibration frequencies of (mono, di, tri and tetra) rings layer for (7,7) armchair SWCNT applying G03 program by PM3 method.

Arm Chair (SWCNT)	C-H _{str.}	C- C _{astr}	C=C _{cstr.}	C--C _{cstr}
Mono-rings layer C ₅₆ H ₂₈	3029.38 Eu	1625.08 Eu	1809.50 Eu	1593.30 Eu
Di-rings layer C ₄₈ H ₂₈	3029.38 Eu	1765.19 Eu	1786.90 Eu	1634.17 Eu
Tri-rings layer C ₁₁₂ H ₂₈	3032.11 Eu	1545.80 Eu	1710.90 Bu	1796.50 Eu
Tetra-rings layer C ₁₄₀ H ₂₈	3029.11 Eu	1602.70 Eu	1793.22 Eu	1796.50 Eu

Finally, the calculations show that, similar to the carbon nanotubes [39-40], the charge densities are mainly concentrated at the circumferential carbon and hydrogen atoms of mono and multi-rings layer SWCNT, parallel with their physical properties for

electrical conductivity. The axial carbon atoms have diminishing charges from outer to centre. The H atoms are positively charged, the C atoms are of the negative charge, Fig. 6.

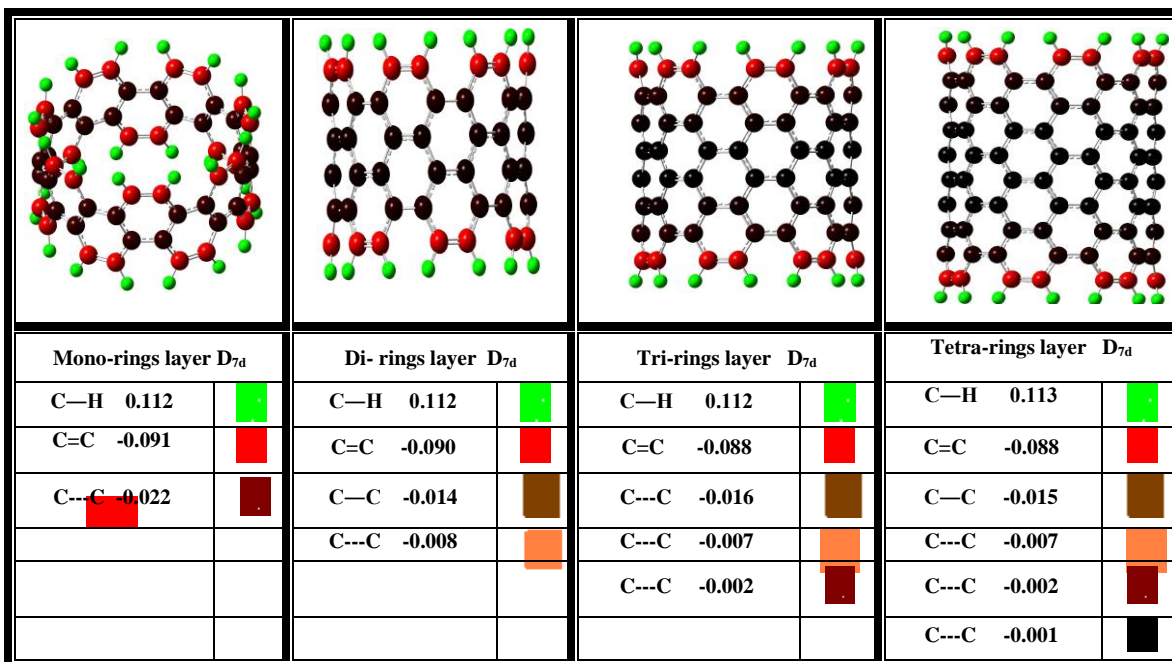


Fig.6: Distribution of charge density at the atoms for (mono, di, tri and tetra) rings layer of (7,7) armchair SWCNT as calculated applying G03 program.

Note. All calculations were done applying the G03 program of Pople et al. [24].

Conclusions

Quantum mechanics semi-empirical (PM3) calculations were carried out with Gaussian 03 program, for investigated the result In order to calculate the vibration frequencies modes that active in infrared. Comparison of the Vibration modes which were active in IR for the three molecules ($C_{56}H_{28}$, $C_{84}H_{28}$, $C_{112}H_{28}$, $C_{140}H_{28}$).

Vibrational motion of atoms decreases the electrical conductivity of nanotubes and limits the performance of nanotransistors and other electronic devices based on them. Similarly, the stiffness of a nanotube and its capability for transporting heat is reduced with increasing defect density[41,42]. They allow a comparative view of the charge density at the carbon atom too. The calculations show that, the charge densities are mainly concentrated at the hydrogen atoms (positively charged) and at the outer circumferential carbon atoms (negatively charge). The axial carbon atoms and the circumferential carbon atoms have diminishing charges from outer to the centre of the CNTs. They can have metallic properties (armchair, some zigzag) depending on their structure. CNTs are much better conductors of electricity than copper nanowires because there is less scatter of electrons. Nanotubes are exceptional ballistic conductors, which can be used in a variety of nanodevices that can operate at room temperature. Nanotubes possess extraordinary mechanical properties and are among the strongest materials known.

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